

ANNEX TO THE INTERNATIONAL SEARCH REPORT ON

INTERNATIONAL APPLICATION NO. PCT/US 86/01511 (SA 14129)

This Annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on 14/11/86

The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

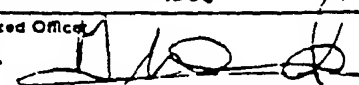
Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A- 0121974	17/10/84	AU-A- 2677284 NL-A- 8301284 JP-A- 59230053	18/10/84 01/11/84 24/12/84
US-A- 4391950	05/07/83	None	
DE-A- 3117514	02/12/82	None	
US-A- 3652710	28/03/72	None	
EP-A- 0024120	25/02/81	JP-A- 56026913 US-A- 4315086 CA-A- 1155583 JP-A- 56047432 JP-A- 57010642	16/03/81 09/02/82 18/10/83 30/04/81 20/01/82
US-A- 4600741	15/07/86	WO-A- 8602086 EP-A- 0195815	10/04/86 01/10/86

For more details about this annex :
see Official Journal of the European Patent Office, No. 12/82

INTERNATIONAL SEARCH REPORT

International Application No PCT/US 86/01511

I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all) *		
According to International Patent Classification (IPC) or to both National Classification and IPC		
IPC ⁴ : C 08 G 65/48; C 08 L 71/04; C 08 L 77/00; C 08 L 67/02		
II. FIELDS SEARCHED		
Minimum Documentation Searched ⁷		
Classification System	Classification Symbols	
IPC ⁴	C 08 G; C 08 L	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched *		
III. DOCUMENTS CONSIDERED TO BE RELEVANT ³		
Category *	Citation of Document, ¹¹ with Indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³
X	EP, A, 0121974 (GEN. ELECTRIC) 17 October 1984, see claims; page 2, paragraph 3; page 3, paragraph 5	1-10
X	US, A, 4391950 (G.D. COOPER) 5 July 1983, see claims; column 3, lines 39-44	1-10
A	DE, A, 3117514 (BASF) 2 December 1982, see claims; pages 4,5	1
A	US, A, 3652710 (F.F. HOLUB et al.) 28 March 1972, see claims	1,11
A	EP, A, 0024120 (SUMITOMO) 25 February 1981, see claims	1,11
P,X	US, A, 4600741 (D.F. AYCOCK et al.) 15 July 1986, see claims	1,11

<p>* Special categories of cited documents: ¹⁰</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&" document member of the same patent family</p>		
IV. CERTIFICATION		
Date of the Actual Completion of the International Search	Date of Mailing of this International Search Report	
3rd November 1986	05 DEC 1986	
International Searching Authority	Signature of Authorized Officer	
EUROPEAN PATENT OFFICE	M. VAN MOL 	

Form PCT/ISA/210 (second sheet) (January 1985)

-28-

17. A method according to claim 15 wherein the functionalizing compound is maleic anhydride.

18. A method according to claim 15 wherein the functionalizing compound is fumaric acid.

19. A method according to claim 15 wherein the impact modifying resin is a diblock or triblock copolymer of styrene and at least one olefin or conjugated diene, any conjugated diene block being non-hydrogenated or partially
5 or entirely hydrogenated, and is present in an amount up to about 50 parts by weight per 100 parts of polyphenylene ether.

20. A method according to claim 15 wherein the temperature of step A is within the range of about 230-390°C.

21. A method according to claim 15 wherein the polyamide is nylon-66.

WHP/tg

-27-

one of styrene homopolymers, high impact polystyrenes and impact modifying resins for polyphenylene ether-polyamide compositions.

12. A method according to claim 11 wherein the polyamide is nylon-6 or nylon-66 and the polyphenylene ether comprises a plurality of structural units having the formula



and in each of said units independently, each Q¹ is independently halogen, primary or secondary lower alkyl, phenyl, haloalkyl, aminoalkyl, hydrocarbonoxy, or halohydrocarbonoxy wherein at least two carbon atoms separate the halogen and oxygen atoms; and each Q² is independently hydrogen, halogen, primary or secondary lower alkyl, phenyl, haloalkyl, hydrocarbonoxy or halohydrocarbonoxy as defined for Q¹.

13. A method according to claim 12 wherein an impact modifying resin is blended into the composition during step A.

14. A method according to claim 13 wherein the functionalizing compound is maleic anhydride, a maleimide or fumaric acid.

15. A method according to claim 13 wherein the polyphenylene ether is a poly(2,6-dimethyl-1,4-phenylene ether).

16. A method according to claim 15 wherein up to about 90% by weight of the total polyphenylene oxide blended into the composition is unfunctionalized.

-26-

5. A method according to claim 3 wherein the functionalizing compound is maleic anhydride, a maleimide or fumaric acid.

6. A method according to claim 3 wherein the polyphenylene ether is a poly(2,6-dimethyl-1,4-phenylene ether).

7. A method according to claim 6 wherein the functionalizing compound is maleic anhydride.

8. A method according to claim 6 wherein the functionalizing compound is fumaric acid.

9. A method according to claim 6 wherein the impact modifying resin is a diblock or triblock copolymer of styrene and at least one olefin or conjugated diene, any conjugated diene block being non-hydrogenated or partially or entirely hydrogenated, and is present in an amount up to about 50 parts by weight per 100 parts of polyphenylene ether.

10. A method according to claim 6 wherein the temperature of mixing is within the range of about 230-390°C.

11. A method for preparing a polyphenylene ether-polyamide composition which comprises:

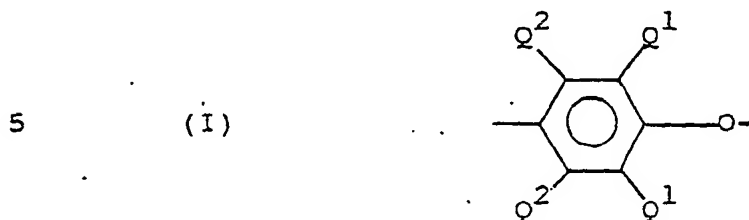
- (A) mixing at least a portion of said polyphenylene ether in the melt, in the absence of free radical initiators, with at least one functionalizing compound having within its molecular structure at least one carboxylic acid, acid anhydride, acid amide, imide, ester, amino or hydroxy group, thereby producing a functionalized polyphenylene ether; and subsequently
- (B) melt-blending said functionalized polyphenylene ether with at least one polyamide; with the proviso that there is also blended into said composition at least

-25-

What is claimed is:

1. A method for functionalizing a polyphenylene ether which comprises mixing said polyphenylene ether in the melt, in the absence of free radical initiators, with at least one functionalizing compound having within its molecular structure at least one carbon-carbon double or triple bond and at least one carboxylic acid, acid anhydride, acid amide, imide, ester, amino or hydroxy group.

2. A method according to claim 1 wherein the polyphenylene ether comprises a plurality of structural units having the formula



and in each of said units independently, each Q^1 is independently halogen, primary or secondary lower alkyl, phenyl, haloalkyl, aminoalkyl, hydrocarbonoxy, or halohydrocarbonoxy wherein at least two carbon atoms separate the halogen and oxygen atoms; and each Q^2 is independently hydrogen, halogen, primary or secondary lower alkyl, phenyl, haloalkyl, hydrocarbonoxy or halohydrocarbonoxy as defined for Q^1 .

3. A method according to claim 2 wherein the mixture also contains an impact modifying resin for polyphenylene ether-polyamide compositions.

4. A method according to claim 2 wherein the mixture also contains a styrene homopolymer or high impact polystyrene.

-24-

Examples 21-25

By a procedure similar to that of Example 11, 58.6 parts of a functionalized polyphenylene ether (containing impact modifier) was extruded with 40 parts of a polyamide-66 and 0.4 part of a hindered phenol stabilizer. The extrudates were molded into test specimens and various physical properties were determined. Comparison was made with controls in which the same ingredients were blended and subjected to a single extrusion. The results are given in Table V, with corresponding controls designated "C".

TABLE V

	Example	21	21C	22	23	23C	24	24C	25
	Functionalized polyphenylene ether	11	--	12	13	--	14	--	15
15	Izod impact strength (notched), joules/m.	577	48	587	443	198	523	150	545
	Tensile strength, MPa.	50	50	52	50	51	51	50	50
	Tensile yield, MPa.	56	50	59	54	54	59	48	57
	Elongation, %	61	25	42	68	77	42	80	40
20	Heat distortion temp., °C	188	184	188	191	189	189	187	190

Again, the improved impact strength resulting from the blending method of this invention is apparent.

-23-

Examples 14-15

Polyphenylene ether-polyamide compositions containing the functionalized polyphenylene ethers of Examples 4-5, in combination with the polyamide of Example 11 and the impact modifier of Examples 1-3, were prepared by tumble mixing the ingredients on a roll mill for 30 minutes and extruding in a Welding Engineers twin screw extruder at 400 rpm. and 288°C, with full vacuum vent. The relative parameters and test results are given in Table IV. The controls were prepared similarly, using unfunctionalized polyphenylene ether and including fumaric acid in the blends.

TABLE IV

	14	15	A	B	C
Polyphenylene ether composition:					
15 Unfunctionalized	--	--	49	49	49
Product of Example 4	59.7	--	--	--	--
Product of Example 5	--	60.5	--	--	--
Polyamide	41	41	41	41	41
Block copolymer	--	--	10	10	10
20 Fumaric acid	--	--	0.15	0.35	0.7
Izod impact strength, (notched), joules/m.	748	721	'50	'50	'50

In these examples, the effect of preparing a functionalized polyphenylene ether according to the method of this invention, in combination with an impact improving polymer, is evident.

Examples 16-20

Blends are prepared by the procedures of Examples 11-15, substituting for the polyamide-66 a similar polyamide-6. Comparable results are obtained.

-22-

Examples 12-13

Polyphenylene ether-polyamide compositions containing the products of Examples 2-3 were prepared by extrusion as described in Example 6. The relative proportions and parameters are given in Table III. The controls are blends prepared similarly, but not containing the functionalized polyphenylene ethers.

TABLE III

	Example		Control			
	12	13	A	B	C	D
Polyphenylene ether:						
Unfunctionalized	--	--	45	45	45	45
Product of Example 2	50.25	--	--	--	--	--
Product of Example 3	--	55.25	--	--	--	--
Polyamide	50	45	45	45	45	45
Block copolymer	--	--	10	10	10	10
Maleic anhydride	--	--	0.25	0.05	0.2	1.0
Thermal stabilizers	0.4	0.4	0.4	--	--	--
Tensile strength upon						
fracture, MPa	70.1	61.2	51.7	53.9	56.3	54.9
Elongation upon						
fracture, %	42	72	45	35	52	81
Izod impact strength						
(notched), joules/m.	220	605	145	157.5	217.5	252.5
Falling dart impact,						
joules	210	217	133	82	215	189

The results in Table III show the improvement in tensile strength resulting from the preparation of a functionalized polyphenylene ether in combination with the impact strength improving agent. For the most part, there is also an improvement in falling dart impact, and, in the case of Example 13, Izod impact strength also.

-21-

Example 11

A blend containing the functionalized polyphenylene ether of Example 1, in combination with unfunctionalized polyphenylene ether, polyamide and impact modifier, was prepared by extrusion on a Werner-Pfleiderer twin screw extruder at 300 rpm. and 285°C.. The polyamide was a pre-dried polyamide-66 having a number average molecular weight of about 20,000 and a moisture content after drying less than 0.2% by weight. The impact modifier was a non-hydrogenated styrene-butadiene-styrene triblock copolymer having a styrene-butadiene ratio of 28:72 and a number average molecular weight of about 80,000.

The relative proportions and parameters are given in Table II, in comparison with a control using unfunctionalized polyphenylene ether and maleic anhydride. Proportions of constituents are in parts by weight.

TABLE II

		Example 6	Control
	Polyphenylene ether:		
20	Unfunctionalized	24.5	49
	Product of Example 1	24.5	--
	Polyamide	41	41
	Block copolymer	10	10
	Maleic anhydride	--	0.25
25	Tensile strength upon fracture, MPa.	59	59
	Elongation upon fracture, %	55	63
	Izod impact strength (notched), joules/m.	332	240
	Falling dart impact, joules	220*	170

*Did not fracture.

The results in Table II show the advantages with respect to impact strength provided by using a functionalized polyphenylene oxide prepared by the method of this invention.

-20-

5-75% by weight polyphenylene ether and about 25-95% polyamide or polyester.

According to the method of this invention for preparing polyphenylene ether-polyamide compositions, there is also blended into said composition at least one of the previously identified impact modifiers. The impact modifier may be introduced at any time during the blending operation, or over an extended time period separate from or concurrent with the addition of other blend constituents.

It is preferred, however, to complete addition of the impact modifier before beginning polyamide addition. More preferably, all of the impact modifier is blended with the polyphenylene ether concurrent with functionalization thereof (i.e., during step A), as described hereinabove.

Blending may be achieved by known methods, typically involving melt blending and extrusion. In one suitable method, the functionalized polyphenylene ether is prepared in a first extruder, preferably in the presence of the impact modifier as previously noted, and subsequently blended in a second extruder with the remaining constituents, including polyamide and any unfunctionalized polyphenylene ether.

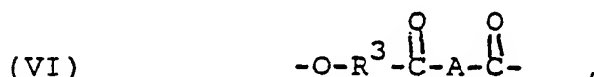
It is also possible to perform the entire blending operation in a lateral-feed extruder, in which it is possible to supply constituents at various points. Under these conditions, the polyphenylene ether, the functionalizing agent and preferably the impact modifier are supplied at the rear of the extruder, and any remaining constituents are supplied at one or more points nearer its outlet.

The effect on the properties of polyphenylene ether-polyamide compositions of the functionalized polyphenylene ethers prepared by the method of this invention is illustrated by the following examples.

- 19 -

Linear polyesters which may be blended with the functionalized polyphenylene ethers include thermoplastic poly(alkylene dicarboxylates) and alicyclic analogs thereof. They typically comprise structural units of the formula

5



wherein R^3 is a saturated divalent aliphatic or alicyclic hydrocarbon radical containing about 2-10 and usually about 2-6 carbon atoms and A is a divalent aromatic radical containing about 6-20 carbon atoms. They are ordinarily prepared by the reaction of at least one diol such as ethylene glycol, 1,4-butanediol or 1,4-cyclohexanedimethanol with at least one aromatic dicarboxylic acid such as isophthalic or terephthalic acid, or lower alkyl ester thereof.

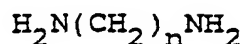
The polyalkylene terephthalates, particularly polyethylene and polybutylene terephthalate and especially the latter, are preferred. Such polyesters are known in the art as illustrated by the following patents:

20	2,465,319	3,047,539
	2,720,502	3,671,487
	2,727,881	3,953,394
	2,822,348	4,128,526.

The polyphenylene ether which is blended with the polyamide or polyester may be solely functionalized polyphenylene ether prepared by the method of this invention. However, it is also contemplated to include both functionalized and unfunctionalized polyphenylene ether in the blend, the latter being present in an amount up to about 90% by weight of total polyphenylene ether. The polyphenylene ether-polyamide or polyester blend generally contains about

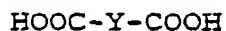
-18-

alkyl, aryl and alkaryl diamines. Such diamines include, for example, those represented by the general formula



wherein n is an integer of from 2 to 16. Illustrative
5 diamines are trimethylenediamine, tetramethylenediamine, pentamethylenediamine, octamethylenediamine, hexamethylenediamine (which is often preferred), trimethylhexamethylenediamine, m-phenylenediamine and m-xylylenediamine.

The dicarboxylic acids may be represented by the
10 formula



wherein Y is a divalent aliphatic or aromatic group containing at least 2 carbon atoms. Examples of aliphatic acids are sebacic acid, octadecanedioic acid, suberic acid, glutaric acid, pimelic acid and adipic acid. Aromatic acids,
15 such as isophthalic and terephthalic acids, are preferred.

Typical examples of the polyamides or nylons, as these are often called, include, for example, polyamide-6, 66, 11, 12, 63, 64, 6/10 and 6/12 as well as polyamides from
20 terephthalic acid and/or isophthalic acid and trimethylhexamethylenediamine; from adipic acid and m-xylylenediamines; from adipic acid, azelaic acid and 2,2-bis-(p-aminocyclohexyl)propane and from terephthalic acid and 4,4'-diaminodicyclohexylmethane. Mixtures and/or copolymers of two or
25 more of the foregoing polyamides or prepolymers thereof, respectively, are also within the scope of the present invention. Preferred polyamides are polyamide-6, 66, 11 and 12, most preferably polyamide-66.

-17-

(B) melt-blending said functionalized polyphenylene ether with at least one polyamide; with the proviso that there is also blended into said composition at least one impact modifying resin for polyphenylene ether-polyamide compositions.

Suitable polyamides may be made by any known method, including the polymerization of a monoamino-mono-carboxylic acid or a lactam thereof having at least 2 carbon atoms between the amino and carboxylic acid group, of substantially equimolar proportions of a diamine which contains at least 2 carbon atoms between the amino groups and a dicarboxylic acid, or of a monoaminocarboxylic acid or a lactam thereof as defined above together with substantially equimolar proportions of a diamine and a dicarboxylic acid. (The term "substantially equimolar" proportions includes both strictly equimolar proportions and slight departures therefrom which are involved in conventional techniques for stabilizing the viscosity of the resultant polyamides.) The dicarboxylic acid may be used in the form of a functional derivative thereof, for example, an ester or acid chloride.

Examples of the aforementioned monoamino-monocarboxylic acids or lactams thereof which are useful in preparing the polyamides include those compounds containing from 2 to 16 carbon atoms between the amino and carboxylic acid groups, said carbon atoms forming a ring with the -CO-NH-group in the case of a lactam. As particular examples of aminocarboxylic acids and lactams there may be mentioned ϵ -aminocaproic acid, butyrolactam, pivalolactam, ϵ -caprolactam, capryllactam, enantholactam, undecanolactam, dodecanolactam and 3- and 4-aminobenzoic acids.

Diamines suitable for use in the preparation of the polyamides include the straight chain and branched chain

Examples 6-10

By a procedure similar to that of Examples 1-3, functionalized polyphenylene ethers were prepared from 46 parts of polyphenylene ether, 0.6 part of fumaric acid and 12 parts of the following impact modifiers:

Example 6 - 75% of a triblock copolymer similar to that of Example 1 but with a somewhat higher molecular weight, 25% of an EPDM rubber.

Example 7 - 100% of the triblock copolymer of Example 6.

Example 8 - 75% of the triblock copolymer of Example 1, 25% of the EPDM rubber of Example 6.

Example 9 - 75% of a partially hydrogenated styrene-isoprene (26:74) diblock copolymer, 25% of the EPDM rubber of Example 6.

Example 10 - 100% of the diblock copolymer of Example 9.

As previously noted, the functionalized polyphenylene ethers prepared by the method of this invention are useful in the preparation of blends of polyphenylene ethers with polyamides and linear polyesters. In particular, another aspect of the invention is a method for preparing a polyphenylene ether-polyamide composition which comprises:

(A) mixing at least a portion of said polyphenylene ether in the melt, in the absence of free radical initiators, with at least one functionalizing compound as described hereinabove, thereby producing a functionalized polyphenylene ether; and subsequently

-15-

Examples 1-3

Mixtures of polyphenylene ether, maleic anhydride and, in Examples 2 and 3, a triblock copolymer were extruded in a Werner-Pfleiderer twin screw extruder at 300 rpm. and 285°C. The triblock copolymer used was a partially hydrogenated styrene-butadiene-styrene copolymer having a styrene-butadiene ratio of 27:73 and a number average molecular weight of about 74,000. The proportions of ingredients are given in Table I.

TABLE I

	Example	Parts		
		1	2	3
	Polyphenylene ether	100	90	80
	Maleic anhydride	1	0.5	0.5
15	Block copolymer	-	10	20

The products were the desired functionalized polyphenylene ethers.

Examples 4-5

Mixtures of 49 parts of polyphenylene ether, 10 parts of the triblock copolymer of Examples 1-3 and various proportions of fumaric acid were tumble mixed on a roll mill for 30 minutes and extruded in a Welding Engineers twin screw extruder at 400 rpm. and 288°C, with full vacuum vent, to yield the desired functionalized polyphenylene ethers.

The proportions of fumaric acid used were as follows:

Example 4 - 0.7 part

Example 5 - 1.5 parts.

-14-

Also useful are the core-shell copolymers wherein an interpenetrating network of the resins employed characterizes the interface between the core and shell. Especially preferred in this regard are the ASA type copolymers available from General Electric Company and sold as GELOY™ resin and described in U.S. Patent 3,944,631.

In addition, there may be employed the above-described polymers and copolymers having copolymerized therewith or grafted thereon monomers having functional groups and/or polar or active groups. Finally, other suitable impact modifiers include Thiokol rubber, polysulfide rubber, polyurethane rubber, polyether rubber (e.g., polypropylene oxide), epichlorohydrin rubber, ethylene-propylene rubber, thermoplastic polyester elastomers and thermoplastic ether-ester elastomers.

The proportion of impact modifier or other resinous material is subject to wide variation. Generally, about 1-150 parts by weight of styrene homopolymer or HIPS is employed per 100 parts of polyphenylene ether. Impact modifiers such as diblock or triblock copolymers are usually present in an amount up to about 50 parts per 100 parts of polyphenylene ether.

The preparation of the functionalized polyphenylene ethers of this invention is illustrated by the following examples. All parts and percentages are by weight. The polyphenylene ether used in these examples was a poly-(2,6-dimethyl-1,4-phenylene ether) having a number average molecular weight of about 20,000 and an intrinsic viscosity in chloroform at 25°C of 0.48-0.49 dl./g.

-13-

(α -methylstyrene) and
poly(α -methylstyrene)-polyisoprene-poly-
(α -methylstyrene).

Particularly preferred triblock copolymers are available
5 commercially as CARIFLEX®, KRATON D® and KRATON G® from
Shell.

Another class of impact modifiers is derived from
conjugated dienes. While many copolymers containing conju-
gated dienes have been discussed above, additional conjugat-
10 ed diene modifier resins include, for example, homopolymers
and copolymers of one or more conjugated dienes including,
for example, polybutadiene, butadiene-styrene copolymers,
isoprene-isobutylene copolymers, chlorobutadiene polymers,
butadiene-acrylonitrile copolymers, polyisoprene, and the
15 like. Ethylene-propylene-diene monomer rubbers may also be
used. These EPDM's are typified as comprising predominantly
ethylene units, a moderate amount of propylene units and up
to about 20 mole percent of non-conjugated diene monomer
units. Many such EPDM's and processes for the production
20 thereof are disclosed in U.S. Patents 2,933,480; 3,000,866;
3,407,158; 3,093,621 and 3,379,701, incorporated herein by
reference.

Other suitable impact modifiers are the core-shell
type graft copolymers. In general, these have a predomi-
25 nantly conjugated diene rubbery core or a predominantly
cross-linked acrylate rubbery core and one or more shells
polymerized thereon and derived from monoalkenylarene and/or
acrylic monomers alone or, preferably, in combination with
other vinyl monomers. Such core-shell copolymers are widely
30 available commercially, for example, from Rohm and Haas
Company under the trade names KM-611, KM-653 and KM-330, and
are described in U.S. Patents 3,808,180; 4,034,013;
4,096,202; 4,180,494 and 4,292,233.

-12-

A particularly useful class of impact modifiers are those derived from the vinyl aromatic monomers. These include AB and ABA type block and radial block copolymers and vinyl aromatic conjugated diene core-shell graft copolymers.

An especially preferred subclass of vinyl aromatic monomer-derived resins is the block copolymers comprising monoalkenyl arene (usually styrene) blocks and conjugated diene (e.g., butadiene or isoprene) or olefin (e.g., ethylene-propylene, ethylene-butylene) blocks and represented as AB and ABA block copolymers. The conjugated diene blocks may be partially or entirely hydrogenated, whereupon the properties are similar to the olefin block copolymers.

Suitable AB type block copolymers are disclosed in, for example, U.S. Patents 3,078,254; 3,402,159; 3,297,793; 3,265,765 and 3,594,452 and UK Patent 1,264,741, all incorporated herein by reference. Exemplary of typical species of AB block copolymers there may be given:

polystyrene-polybutadiene (SBR)
polystyrene-polyisoprene and
poly(α -methylstyrene)-polybutadiene.

Such AB block copolymers are available commercially from a number of sources, including Phillips Petroleum under the trademark SOLPRENE.

Additionally, ABA triblock copolymers and processes for their production as well as hydrogenation, if desired, are disclosed in U.S. Patents 3,149,182; 3,231,635; 3,462,162; 3,287,333; 3,595,942; 3,694,523 and 3,842,029, all incorporated herein by reference.

Examples of triblock copolymers include:
polystyrene-polybutadiene-polystyrene (SBS),
polystyrene-polyisoprene-polystyrene (SIS),
poly(α -methylstyrene)-polybutadiene-poly-

-11-

vinyltoluene, random copolymers of styrene with one or more monomers illustrated by acrylonitrile, butadiene, α -methylstyrene, ethylvinylbenzene, divinylbenzene and maleic anhydride, and rubber-modified polystyrenes comprising blends and grafts, wherein the rubber is a polybutadiene or a rubbery copolymer of about 98-70% styrene and about 2-30% diene monomer. These rubber-modified polystyrenes include high impact polystyrene, or HIPS.

Impact modifiers for polyphenylene ether-polyamide compositions are well known in the art. They are typically derived from one or more monomers selected from the group consisting of olefins, vinyl aromatic monomers, acrylic and alkylacrylic acids and their ester derivatives as well as conjugated dienes. Especially preferred impact modifiers are the rubbery high-molecular weight materials including natural and synthetic polymeric materials showing elasticity at room temperature. They include both homopolymers and copolymers, including random, block, radial block, graft and core-shell copolymers as well as combinations thereof.

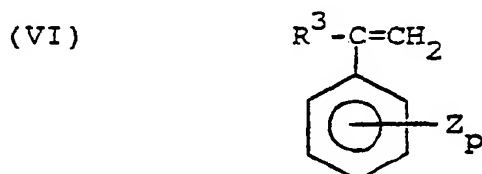
Polyolefins or olefin-based copolymers employable in the invention include low density polyethylene, high density polyethylene, linear low density polyethylene, isotactic polypropylene, poly(1-butene), poly(4-methyl-1-pentene), propylene-ethylene copolymers and the like. Additional olefin copolymers include copolymers of one or more α -olefins, particularly ethylene, with copolymerizable monomers including, for example, vinyl acetate, acrylic acids and alkylacrylic acids as well as the ester derivatives thereof including, for example, ethylene-acrylic acid, ethyl acrylate, methacrylic acid, methyl methacrylate and the like. Also suitable are the ionomer resins, which may be wholly or partially neutralized with metal ions.

-10-

the functionalized polyphenylene ether when combined with polyamides and polyesters. The principal reaction (if any) is more likely a thermally initiated free radical interaction of the carbon-carbon double or triple bonds with the aromatic rings or the substituents thereon, especially the latter, to produce a product which may include single moieties and/or grafted side chains derived from the functionalizing agent. There may also be some degree of amine salt formation by reaction of the acidic groups with amines generated during formation of the quinone methide-type intermediate of formula IV. However, the invention is in no way dependent on theory.

As more fully described hereinafter, a principal utility of the functionalized polyphenylene ethers of this invention is in the preparation of polyphenylene ether-polyamide compositions. Such compositions may also contain other resinous materials such as polystyrenes and impact modifying resins, and it is frequently preferred to melt-blend at least one such other resinous material, especially an impact modifier, with the polyphenylene ether and the functionalizing compound.

The term "polystyrene" as used herein includes resins comprising polymers which contain at least 25% by weight of structural units derived from a monomer of the formula



wherein R^3 is hydrogen, lower alkyl or halogen; Z is vinyl, halogen or lower alkyl; and p is from 0 to 5. These resins include homopolymers of styrene, chlorostyrene and

-9-

230-390°C. The fact that some type of interaction takes place in the absence of free radical initiators, and that the product is useful for compatibilizing blends as described hereinafter, is quite unexpected in view of the
5 aforementioned Japanese Kokai 84/66452, which explicitly states by way of comparative test results that such products are of little or no utility for this purpose.

The proportions of polyphenylene ether and functionalizing compound are not critical, provided the functionalizing compound is used in minor proportions compared
10 to the polyphenylene ether. Most often, about 0.01-5.0 parts and preferably about 0.1-3.0 parts of functionalizing agent are present per 100 parts of polyphenylene ether.

The mixing conditions often include extrusion,
15 which may be conveniently effected in a screw-type or similar extruder which supplies a substantial shearing force to the composition. In certain instances, it may be advantageous to vacuum vent the extruder by connecting the vent thereof to a vacuum pump capable of drawing a vacuum of
20 about 20 torr or less. It is also sometimes found advantageous to extrude the mixture more than once, thereby ensuring effective blending.

The precise chemical nature of the functionalization which takes place upon practice of the method of this
25 invention is not known with certainty. It is believed that an esterification reaction occurs to some extent between compounds such as maleic anhydride or fumaric acid and the terminal hydroxy groups on the polyphenylene ether, particularly on the 4-hydroxybiphenyl end groups of formula III.
30 Another possible reaction is of the Diels-Alder type between maleic acid, maleimide, fumaric acid or the like and the quinone methide of formula IV. However, said reactions are not presently believed to account for the high activity of

-8-

90% by weight of the polymer, contain end groups having one or frequently both of formulas II and III. It should be understood, however, that other end groups may be present and that the invention in its broadest sense may not be dependent on the molecular structures of the polyphenylene ether end groups.

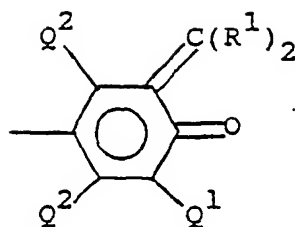
It will be apparent to those skilled in the art from the foregoing that the polyphenylene ethers contemplated for use in the present invention include all those presently known, irrespective of variations in structural units or ancillary chemical features.

The functionalizing compound employed according to the present invention, as previously described, contains one or more carbon-carbon double and/or triple bonds in combination with various functional groups including amino, hydroxy, carboxylic acid and carboxylic acid derivative groups. Illustrative compounds of this type are maleic acid, fumaric acid, maleic anhydride, maleimides such as N-phenylmaleimide and 1,4-phenylene-bis-methylene- α,α' -bismaleimide, maleic hydrazide, methylnadic anhydride, fatty oils (e.g., soybean oil, tung oil, linseed oil, sesame oil), unsaturated carboxylic acids such as acrylic acid, crotonic acid, methacrylic acid and oleic acid, unsaturated alcohols such as allyl alcohol and crotyl alcohol and unsaturated amines such as allylamine. The preferred compounds are maleic anhydride, maleimides and fumaric acid, particularly the latter by reason of its availability, relative non-toxicity and low activity as a skin and eye irritant.

The functionalized polyphenylene ethers may be conveniently prepared by merely blending the two reagents under conditions adapted for the formation of an intimate blend, and at temperatures high enough to prepare a melt. Typical temperatures are within the range of about

-7-

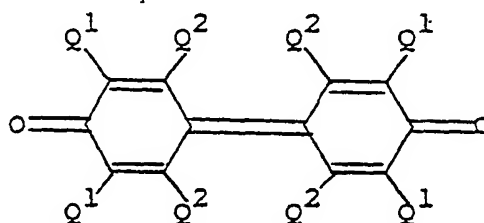
(IV)



with numerous beneficial effects often including an increase
 5 in impact strength and compatibilization with other blend
 components. Reference is made to U.S. Patents 4,054,553,
 4,092,294, 4,477,649, 4,477,651 and 4,517,341, the disclo-
 sures of which are incorporated by reference herein.

Polymers with 4-hydroxybiphenyl end groups of
 10 formula III are typically obtained from reaction mixtures in
 which a by-product diphenoquinone of the formula

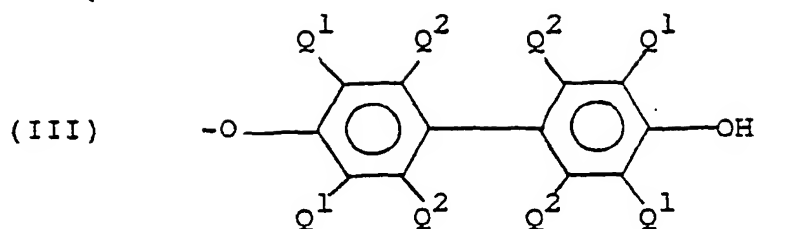
(V)



15 is present, especially in a copper-halide-secondary or
 tertiary amine system. In this regard, the disclosure of
 U.S. Patent 4,477,649 is again pertinent as are those of
 U.S. 4,234,706 and 4,482,697, which are also incorporated by
 reference herein. In mixtures of this type, the dipheno-
 20 quinone is ultimately incorporated into the polymer in
 substantial proportions, largely as an end group.

In many polyphenylene ethers obtained under the
 above-described conditions, a substantial proportion of the
 polymer molecules, typically constituting as much as about

-6-



wherein Q^1 and Q^2 are as previously defined; each R^1 is
 5 independently hydrogen or alkyl, with the proviso that the
 total number of carbon atoms in both R^1 radicals is 6 or
 less; and each R^2 is independently hydrogen or a C_{1-6} pri-
 mary alkyl radical. Preferably, each R^1 is hydrogen and
 each R^2 is alkyl, especially methyl or n-butyl.

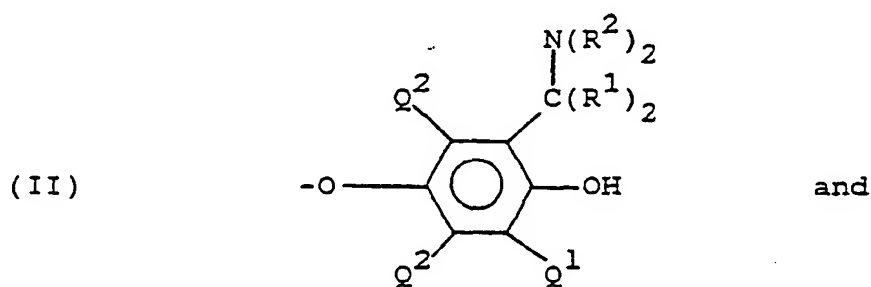
10. Polymers containing the aminoalkyl-substituted end
 groups of formula II may be obtained by incorporating an
 appropriate primary or secondary monoamine as one of the
 constituents of the oxidative coupling reaction mixture,
 especially when a copper- or manganese-containing catalyst
 15 is used. Such amines, especially the dialkylamines and
 preferably di-n-butylamine and dimethylamine, frequently
 become chemically bound to the polyphenylene ether, most
 often by replacing one of the α -hydrogen atoms on one or
 more Q^1 radicals. The principal site of reaction is the Q^1
 20 radical adjacent to the hydroxy group on the terminal unit
 of the polymer chain. During further processing and/or
 blending, the aminoalkyl-substituted end groups may undergo
 various reactions, probably involving a quinone methide-type
 intermediate of the formula

-5-

are disclosed, for example, in U.S. Patents 3,306,874, 3,306,875, 3,914,266 and 4,028,341. They are usually combinations of cuprous or cupric ions, halide (i.e., chloride, bromide or iodide) ions and at least one amine.

5 Catalyst systems containing manganese compounds constitute a second preferred class. They are generally alkaline systems in which divalent manganese is combined with such anions as halide, alkoxide or phenoxide. Most often, the manganese is present as a complex with one or
10 more complexing and/or chelating agents such as dialkylamines, alkanolamines, alkylenediamines, o-hydroxyaromatic aldehydes, o-hydroxyazo compounds, o-hydroxyoximes (monomeric and polymeric), o-hydroxyaryl oximes and β -diketones. Also useful are known cobalt-containing catalyst systems.
15 Suitable manganese and cobalt-containing catalyst systems for polyphenylene ether preparation are known in the art by reason of disclosure in numerous patents and publications.

Particularly useful polyphenylene ethers for the purposes of this invention are those which comprise molecules having at least one of the end groups of the formulas
20



-4-

ungrafted moieties. Other suitable polymers are the coupled polyphenylene ethers in which the coupling agent is reacted in known manner with the hydroxy groups of two polyphenylene ether chains to produce a higher molecular weight polymer containing the reaction product of the hydroxy groups and the coupling agent. Illustrative coupling agents are low molecular weight polycarbonates, quinones, heterocycles and formals.

The polyphenylene ether generally has a number average molecular weight within the range of about 3,000-40,000 and a weight average molecular weight within the range of about 20,000-80,000, as determined by gel permeation chromatography. Its intrinsic viscosity is most often in the range of about 0.35-0.6 dl./g., as measured in chloroform at 25°C.

The polyphenylene ethers are typically prepared by the oxidative coupling of at least one corresponding monohydroxyaromatic compound. Particularly useful and readily available monohydroxyaromatic compounds are 2,6-xyleneol (wherein each Q^1 is methyl and each Q^2 is hydrogen), whereupon the polymer may be characterized as a poly(2,6-dimethyl-1,4-phenylene ether), and 2,3,6-trimethylphenol (wherein each Q^1 and one Q^2 is methyl and the other Q^2 is hydrogen).

A variety of catalyst systems are known for the preparation of polyphenylene ethers by oxidative coupling. There is no particular limitation as to catalyst choice and any of the known catalysts can be used. For the most part, they contain at least one heavy metal compound such as a copper, manganese or cobalt compound, usually in combination with various other materials.

A first class of preferred catalyst systems consists of those containing a copper compound. Such catalysts

-3-

In each of said units independently, each Q^1 is independently halogen, primary or secondary lower alkyl (i.e., alkyl containing up to 7 carbon atoms), phenyl, haloalkyl, aminoalkyl, hydrocarbonoxy, or halohydrocarbonoxy wherein at least two carbon atoms separate the halogen and oxygen atoms; and each Q^2 is independently hydrogen, halogen, primary or secondary lower alkyl, phenyl, haloalkyl, hydrocarbonoxy or halohydrocarbonoxy as defined for Q^1 . Examples of suitable primary lower alkyl groups are methyl, ethyl, n-propyl, n-butyl, isobutyl, n-amyl, isoamyl, 2-methylbutyl, n-hexyl, 2,3-dimethylbutyl, 2-, 3- or 4-methylpentyl and the corresponding heptyl groups. Examples of secondary lower alkyl groups are isopropyl, sec-butyl and 3-pentyl. Preferably, any alkyl radicals are straight chain rather than branched. Most often, each Q^1 is alkyl or phenyl, especially C_{1-4} alkyl, and each Q^2 is hydrogen. Suitable polyphenylene ethers are disclosed in a large number of patents.

Both homopolymer and copolymer polyphenylene ethers are included. Suitable homopolymers are those containing, for example, 2,6-dimethyl-1,4-phenylene ether units. Suitable copolymers include random copolymers containing such units in combination with (for example) 2,3,6-trimethyl-1,4-phenylene ether units. Many suitable random copolymers, as well as homopolymers, are disclosed in the patent literature.

Also included are polyphenylene ethers containing moieties which modify properties such as molecular weight, melt viscosity and/or impact strength. Such polymers are described in the patent literature and may be prepared by grafting onto the polyphenylene ether in known manner such vinyl monomers as acrylonitrile and vinylaromatic compounds (e.g., styrene), or such polymers as polystyrenes and elastomers. The product typically contains both grafted and

-2-

which form compatible compositions when blended with such polymers as polyamides and linear polyesters.

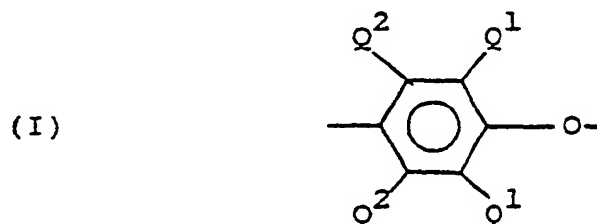
5 A further object is to provide polyphenylene ether-polyamide compositions having high impact strength and solvent resistance.

Other objects will in part be obvious and will in part appear hereinafter.

10 In one of its aspects, the present invention is directed to a method for functionalizing a polyphenylene ether which comprises mixing said polyphenylene ether in the melt, in the absence of free radical initiators, with at least one functionalizing compound having within its molecular structure at least one carbon-carbon double or triple bond and at least one carboxylic acid, acid anhydride, acid
15 amide, imide, ester, amino or hydroxy group.

The polyphenylene ethers used in the present invention are a well known class of polymers. They are widely used in industry, especially as engineering plastics in applications requiring toughness and heat resistance.
20 Since their discovery, they have given rise to numerous variations and modifications all of which are applicable to the present invention, including but not limited to those described hereinafter.

25 The polyphenylene ethers comprise a plurality of structural units having the formula



-1-

FUNCTIONALIZED POLYPHENYLENE ETHERS AND
METHODS OF PREPARATION AND USE

This invention relates to the functionalization of polyphenylene ethers, and to uses for the polyphenylene ethers so functionalized.

Various compositions comprising polyphenylene ethers (also known as polyphenylene oxides) and polyamides or linear polyesters, illustrated by Nylon-6, Nylon-66, poly(ethylene terephthalate) and poly(butylene terephthalate), are of interest because of their high solvent resistance and potentially high impact strength. However, such compositions typically undergo phase separation and delamination because of the presence therein of large, incompletely dispersed polyphenylene ether particles and the lack of phase interaction between the two resin phases.

According to U.S. Patent 4,315,086 and European patent application 24,120, compositions comprising polyphenylene ethers and polyamides, having high impact strength and solvent resistance, may be prepared by melt blending the two polymers with various olefinic or acetylenic carboxylic acids, functional derivatives thereof or other functionalized compounds, and optionally with impact modifiers such as high impact polystyrenes and partially hydrogenated styrene-butadiene block copolymers. Japanese Kokai 84/66452 describes similar compositions prepared by using a polyphenylene ether which has been pretreated with a similar olefinic compound in the mandatory presence of a free radical initiator. There is, however, continuing interest in preparing compositions of this type having still higher impact strengths and other improvements in properties.

A principal object of the present invention, therefore, is to provide functionalized polyphenylene ethers

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GA	Gabon	MR	Mauritania
AU	Australia	GB	United Kingdom	MW	Malawi
BB	Barbados	HU	Hungary	NL	Netherlands
BE	Belgium	IT	Italy	NO	Norway
BG	Bulgaria	JP	Japan	RO	Romania
BR	Brazil	KP	Democratic People's Republic of Korea	SD	Sudan
CF	Central African Republic	KR	Republic of Korea	SE	Sweden
CG	Congo	LI	Licchtenstein	SN	Senegal
CH	Switzerland	LK	Sri Lanka	SU	Soviet Union
CM	Cameroon	LU	Luxembourg	TD	Chad
DE	Germany, Federal Republic of	MC	Monaco	TG	Togo
DK	Denmark	MG	Madagascar	US	United States of America
FI	Finland	ML	Mali		
FR	France				

PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION
International Bureau



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification⁴ : C08G 65/48, C08L 71/04, 77/00 C08L 67/02	A1	(11) International Publication Number: WO 87/ 00540 (43) International Publication Date: 29 January 1987 (29.01.87)
(21) International Application Number: PCT/US86/01511 (22) International Filing Date: 23 July 1986 (23.07.86) (31) Priority Application Number: 8502116 (32) Priority Date: 24 July 1985 (24.07.85) (33) Priority Country: NL (71) Applicant: GENERAL ELECTRIC COMPANY [US/ US]; 1 River Road, Schenectady, NY 12305 (US). (72) Inventors: VAN DER MEER, Roelof ; Jan van Goyen- straat 21, NL-4625 AZ Bergen op Zoom (NL). YATES, John, Bennie, III ; 11 Elm Place, Clifton Park, NY 12065 (US). (74) Agents: KING, Arthur, M. et al.; International Patent Operation, General Electric Company, 570 Lexington Avenue, New York, NY 10022-6853 (US).		(81) Designated States: AU, BR, DE (European patent), FR (European patent), GB (European patent), IT (Euro- pean patent), JP, KR, NL (European patent). Published <i>With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
(54) Title: FUNCTIONALIZED POLYPHENYLENE ETHERS AND METHODS OF PREPARATION AND USE (57) Abstract Functionalized polyphenylene ethers are prepared by mixing with various functionalized olefinic or acetylenic com- pounds in the melt, in the absence of free radical initiator. They are useful in the preparation of compositions comprising polyphenylene ethers and polyamides or polyesters.		